

## Optical limiting and thermal lensing studies in C<sub>60</sub>

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Optical limiting and thermo-optic properties of C<sub>60</sub> in toluene are studied using 532 nm, 9 ns pulses from a frequency-doubled Nd:YAG laser. Optical limiting studies in these fullerene molecules lead to the conclusion that reverse saturable absorption is the major mechanism for limiting properties in these molecules. Thermal lensing measurements are also performed in fullerene solutions. The quadratic dependence of thermal lens signal on incident energy confirms that enhanced optical absorption by the sample via excited triplet state absorption may play a leading role in the limiting property. © 1999 American Institute of Physics. [S0021-8979(99)01715-6]

### I. INTRODUCTION

The story of the discovery of C<sub>60</sub>, otherwise known as fullerene, started in 1985 when Kroto and co-workers<sup>1</sup> evidenced a self-assembled spontaneous C<sub>60</sub> molecule from a hot nucleating carbon plasma. They exploited pulsed laser ablation of graphite in ambient helium as a reliable method for the production of C<sub>60</sub>. Here, a pulsed laser beam is directed onto the surface of a rotating graphite disk. The power density delivered by the laser beam on to the target is sufficient to generate a plasma of carbon vapor over the irradiated spot in which temperatures over 10 000 °C are readily available.<sup>2,3</sup> As the carbon atoms and small radicals are cooled in the ambient carrier gas (usually He) clustering reactions occur.<sup>4</sup> The detailed nature of the carbon clusters can be studied using time-of-flight mass spectroscopy<sup>5,6</sup> and laser spectroscopy.<sup>7,8</sup> The technique developed by Krätschmer *et al.*<sup>9</sup> for preparing and isolating macroscopic quantities of C<sub>60</sub> opened the way for exploring the molecular and bulk properties of these novel species. Fullerenes and their derivatives possess a number of potentially useful physical, biological and chemical properties.

The optical properties of fullerenes have been reported by numerous investigators.<sup>10–17</sup> Photophysical and excited state kinetic properties of fullerenes, including fluorescence have been studied at low temperatures<sup>11,12</sup> as well as at room temperature.<sup>13</sup> Because of the very high rate for intersystem crossing to the excited triplet state at room temperature, the fluorescence emission spectra of these molecules are very weak with an extremely low fluorescent quantum yield of  $\approx 1.45 \times 10^{-4}$  for C<sub>60</sub> at room temperature.<sup>14</sup> The small singlet-triplet splitting, the very low value of fluorescent rate constant and expected large spin-orbital interaction in these nearly spherical molecules indicate the occurrence of intersystem crossing as a dominant process. Early literature reports<sup>11</sup> revealed that C<sub>60</sub> had a higher excited state absorption cross section than the ground state absorption cross section over the complete visible spectrum. This information

implied that the fullerenes are reverse saturable absorption (RSA) materials and may have application to optical limiting for sensor protection.<sup>15</sup> But few authors reported that apart from RSA some other nonlinear mechanism like nonlinear refraction, nonlinear scattering, two photon absorption, etc. also play a major role in the optical limiting properties of these molecules.<sup>16–19</sup>

In this article, the optical limiting properties of C<sub>60</sub> in toluene are discussed along with its thermo-optic properties. We used thermal lensing method to elucidate any nonlinear mechanisms in these liquids. C<sub>60</sub> used in our experiment was prepared by following Krätschmer–Huffman technique<sup>9</sup> and the sample was further purified by employing high performance liquid chromatography. The extract of C<sub>60</sub> has the characteristic color and its electronic absorption spectrum was in agreement with that reported in the literature.<sup>20</sup>

### II. EXPERIMENTAL SETUP

We have carried out measurement of optical limiting properties in C<sub>60</sub> in toluene solutions with quartz cuvettes of path lengths 5 and 10 mm. The cuvette was kept slightly away from the focal spot and a long focal length lens was used so that the spot size inside the cuvette remains constant and was about 500  $\mu\text{m}$  in radius. The input and output laser energies were measured using a laser energy meter. The experimental setup essentially avoids nonlinear refraction as only the linear absorption is important in this case contrary to the measurements done with a point detector to measure the transmitted laser energy.

For thermal lens measurements, a 5 mm sample cell was used. Figure 1 shows the block diagram of the thermal lens spectrophotometer, consisting of a frequency-doubled Q-switched Nd:YAG laser as the heating source and a intensity stabilized He–Ne laser as the probe beam. The sample solution taken in a quartz cuvette having path length 5 mm is placed in the pump beam path. The pump and probe beams are focused onto the sample cell and made to pass collinearly through it using suitable convex lenses and the use of a dichroic mirror. The excitation beam was blocked after the sample cuvette by a filter.

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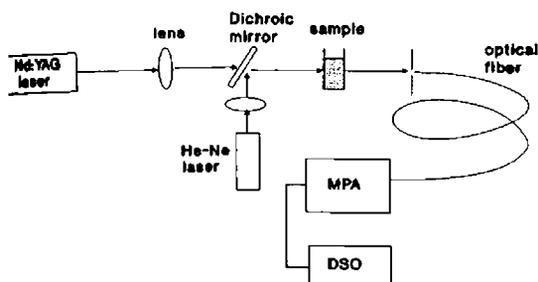


Fig. 1. Pulsed thermal lens setup used in the present experiment (MPA—monochromator—PMT assembly, DSO—digital storage oscilloscope).

As the 532 nm radiation was passed through the sample, molecules absorb some of the incident energy and get excited to higher energy levels. The subsequent deexcitation process can occur radiatively or nonradiatively. It is the nonradiative part that gives rise to thermal lens formation. The resulting refractive index gradient follows the intensity distribution of the exciting pump beam. The TL effect was detected by monitoring the intensity fluctuations in the center of the probe He-Ne laser beam. The position of the cuvette was adjusted to get maximum value for the TL signal. The signal was detected by sampling the intensity of the center portion of the probe beam through a small aperture. In the present work, the intensity of the center portion of the transmitted probe beam was detected by using an optical fiber. The polished tip of a long graded index optical fiber (200  $\mu\text{m}$  core, NA 0.22) serves both as an aperture and as a light guide to direct the probe beam to a monochromator—photomultiplier (PMT) assembly. Using optical fibers to transmit the laser beams makes the thermal lens technique amenable to remote, *in situ* analysis. It also reduces the influence of mode and alignment variations in the probing laser. Here, we used a multimode fiber which minimize alignment procedures and the problems encountered with single mode fibers. The monochromator—photomultiplier assembly tuned to the probe beam wavelength (632.8 nm) provides further filtering of the signal. The TL signal was recorded using a digital storage oscilloscope (100 MHz, Tektronix TDS 220) which provides a complete time domain representation of the signal. The oscilloscope was triggered by a synchronous trigger pulse from the Nd:YAG laser operated at 5 Hz.

### OPTICAL LIMITING IN $C_{60}$

The discovery of optical limiting in fullerenes evoked considerable attention because of its comparatively lower threshold limiting fluence.<sup>15,21,22</sup> Optical limiting property in these new found materials was first reported by Tutt and co-workers<sup>10</sup> using 532 nm, 8 ns pulses. These authors attributed the limiting action mainly to RSA in  $C_{60}$  resulting from the competition between absorption from the photoexcited triplet state compared with that from the ground state. Transient excited state absorption spectroscopy measurements of higher lying singlet and triplet states of these molecules lend support to the hypothesis. Joshi *et al.*<sup>16</sup> did a numerical analysis of a model incorporating the effect of higher excited state

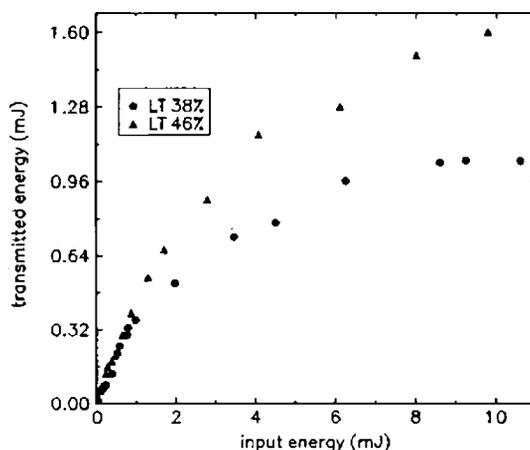


FIG. 2. Variation of transmitted energy with respect to input laser energy for  $C_{60}$  for an optical path length of 5 mm for a linear transmittance (LT) of 46% and 38%.

absorption and found that this model cannot explain their observations. They also observed a reduction in pulse width and steepening of the falling part of the pulse for input energies well into the saturation region and suggests that the multiphoton absorption from the excited state rather than the ground state might be operative. A similar conclusion was reached by McLean *et al.*,<sup>18</sup> who found that their nonlinear transmission measurements with 8 ns and 30 ps, 532 nm pulses departed from the results of a rate equation analysis for input fluences greater than  $1 \text{ J cm}^{-2}$  indicating initiation of some other limiting mechanism. According to them, RSA yields a reasonable explanation of the low fluence behavior in both ns and ps cases, but unusually by large nonradiative relaxation rates for higher lying singlet states must be assumed in the picosecond case. Kost *et al.*<sup>22</sup> reported that nonlinear transmission of  $C_{60}$  in PMMA was in agreement with the RSA mechanism and suggested the stronger limiting in  $C_{60}$  solutions could be due to nonlinear scattering. Even though large attention has been given for the limiting property of these molecules, a clear cut picture for the underlying physics behind this phenomenon has not yet emerged.

The optical limiting is obtained by varying the input energy and measuring the transmitted energy. The variation of transmitted energy with input energy is given in Fig. 2 for different concentrations of  $C_{60}$  in toluene for an optical path length of 5 mm. At very low laser energies, the transmission obeys Beer-Lambert law and the transmitted intensity varies with increasing input energy with a slope equal to  $\exp(-N_0\sigma_1L)$ , where  $\sigma_1$  is the absorption cross section of the ground state,  $N_0$  the number density of the  $C_{60}$  molecules, and  $L$  the cuvette length. At high input intensity, the transmittance decreases with input intensity and we observe an optical limiting property with saturated or clamped output intensity. With solutions showing low intensity linear transmittance (LT) of 46% and 38%, the outgoing laser intensities are saturated at energy levels of 6 and 4.5 mJ, respectively. These data show that as the concentration increases, a reduction in linear transmittance as well as the clamping level is

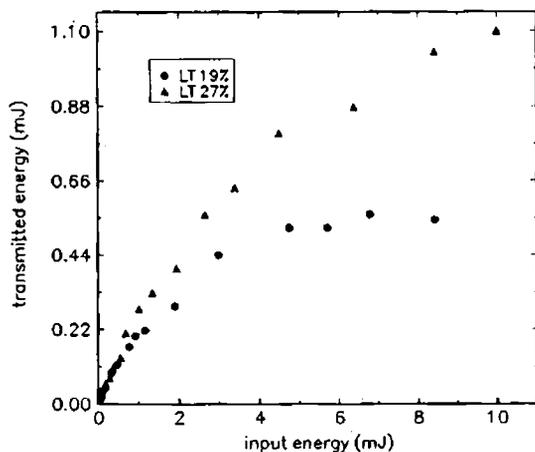


FIG. 3. Variation of output energy vs input energy for an optical pathlength of 10 mm for  $C_{60}$  with a linear transmittance (LT) of 27% and 19%.

observed. It is also noted that the saturated output intensity decreases with increasing length of the optical path. Figure 3 gives the optical limiting properties of these solutions for a 10 mm cell path length. In this experiment, the low intensity linear transmittances of the solutions in a 10 mm path length cell are 27% and 19% with clamping energies at 4.6 and 3 mJ, respectively. This indicates that the number density of  $C_{60}$  in the laser beam is the main factor affecting the clamped level. From the threshold intensity for optical limiting for each sample, it can be seen that the threshold is inversely proportional to the concentration.

Several mechanisms have been proposed for optical limiting in  $C_{60}$  including RSA, nonlinear scattering, multiphoton absorption, cyclic multiphoton absorption, etc. The RSA by a 5 level energy diagram yielded a reasonable explanation for optical limiting in the ns regime of a  $\pi$ -electron conjugated system such as  $C_{60}$  below the input fluence of  $1 J cm^{-2}$ . When a  $C_{60}$  molecule is photoexcited using 532 nm, 9 ns pulse, it gets excited from the ground state  $S_0$  ( $\sigma_g = 1.21 \times 10^{-18} cm^2$ ) into one of the vibrational rotational states of the first excited singlet state  $S_1$ . The molecule relaxes rapidly ( $\approx 1$  ps) into an equilibrium state  $S_1$  which releases 0.33 eV energy. This level relaxes either to the ground state with life time  $\tau_0 = 16$  ns, or to the triplet state  $T_1$  in  $\tau_{s1} = 650$  ps. The reported fluorescence quantum yield<sup>11</sup> for  $C_{60}$  in toluene is  $1.45 \times 10^{-4}$ . But the triplet production yield is found to be nearly unity for 532 nm excitation of  $C_{60}$  molecules. Hence, excited singlet state transfers dominantly to the lower level of the triplet state due to high triplet yield and small  $S-T$  splitting of 9 kcal/mol for  $C_{60}$ . The molecule in the  $S_1$  and  $T_1$  states can be excited to higher  $S_n$  and  $T_n$  states ( $\sigma_{s1} = 8.07 \times 10^{-18} cm^2$  and  $\sigma_{t1} = 5.35 \times 10^{-18} cm^2$ ). Relaxation from these  $S_n$  and  $T_n$  states is in the form of heat and is extremely short ( $< ps$ ). The relaxation of  $T_1$  to  $S_0$  is forbidden and therefore very slow. The triplet state which has long lifetime ( $40 \pm 4 \mu s$  for  $C_{60}$ )<sup>23</sup> and higher absorption cross section than that of the ground state will accumulate significant population and eventually become the dominant absorp-

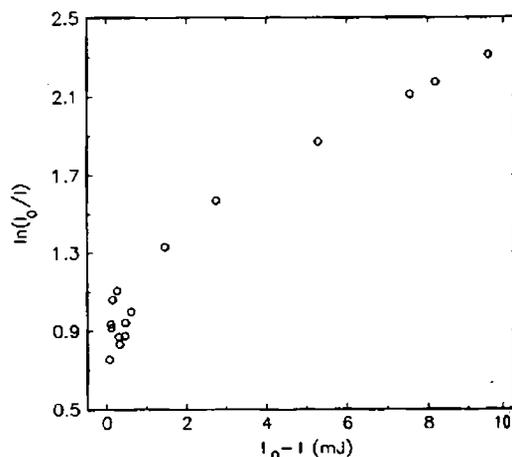


FIG. 4. Plot of  $\ln(I_0/I)$  vs  $(I_0 - I)$  for  $C_{60}$  toluene solution showing approximate linearity.

tion source. At very low laser energies, the transmission creases with increasing input energy with a slope equal to  $\exp(-N_0\sigma_g L)$ , where  $N_0$  the number density of the  $C_{60}$  molecules and  $L$  the optical path length. At higher laser energies taking into account the ground state absorption as well as ESA the transmission varies at a rate  $\exp(N_0\sigma_{s1} + N_1\sigma_{t1})$  where  $N_1$  is the population density for the photoexcited triplet state. It has been found that<sup>11</sup>  $\sigma_2 \approx 2.9\sigma_1$ .

Kojima *et al.*<sup>24</sup> while explaining the optical limiting property of  $\pi$  polyacene based oligomer have shown that in the case of pure RSA the incident laser fluence  $I_0$  and transmitted laser fluence  $I$  obeys the relation,

$$\ln\left(\frac{I_0}{I}\right) = k(I_0 - I) + A_g,$$

where  $k$  is a constant which depends on the absorption cross sections and lifetimes of the ground, excited singlet and excited triplet states and  $A_g$  is the ground state absorbance. The above equation says that the plot of  $\ln(I_0/I)$  vs  $(I_0 - I)$  should be a straight line with slope  $k$  and intercept  $A_g$ . Figure 4 shows such a plot for  $C_{60}$ . The plot of  $\ln(I_0/I)$  vs  $(I_0 - I)$  is approximately linear in the energy range of interest. This is an indication that RSA is one of the limiting mechanisms in the case of  $C_{60}$ .

Even though there are many reports exploring nonlinearities and fluorescence of  $C_{60}$  molecules, its nonradiative relaxations in solutions are not well investigated. Apart from that, there is a controversy existing in the optical limiting mechanism of these molecules. McLean *et al.*<sup>18</sup> suggest various mechanisms other than RSA for the observed optical limiting in  $C_{60}$ . Joshi *et al.*<sup>16</sup> have found that optical limiting in  $C_{60}$  deviates from a theoretical model based on a large excited state absorption cross section and they have attributed the discrepancy to the possible multiphoton absorption (MPA) apart from ESA. The study of MPA in absorbing media can very effectively be carried

ing the transient thermal lens technique.<sup>25,26</sup> We have extended the nonradiative relaxations in C<sub>60</sub> in toluene solution using transient thermal lens technique.

### THERMAL LENSING IN C<sub>60</sub>

Thermal lens spectroscopy is an effective and efficient method to monitor nonradiative relaxations in a medium.<sup>27-30</sup> Very weak optical absorption processes can be studied using this method. Even though a few reports are available<sup>31</sup> in literature on the thermo-optic properties of C<sub>60</sub>, a more detailed study of thermo-optic properties of fullerene molecules has not been undertaken yet. In this section, thermal lensing studies in C<sub>60</sub> are given.

Methods classified as thermal lens spectroscopy are based upon a thermal change in the optical properties of a sample on the absorption of laser energy which leads to a temperature rise in the sample and consequently to the formation of an inhomogeneous spatial profile of the refractive index. The heat released by the nonradiative relaxation processes generates a volume expansion in the sample and a density change within the excitation region. The refractive index change caused by the heat evolution due to the radiationless processes, turns the solution into a divergent lens in most cases, which defocuses the laser beam. The change of irradiance at the probe beam intensity relative to its stationary value  $S$  is given by<sup>27,28</sup>

$$S = \frac{I(t=0) - I(t=\infty)}{I(t=0)} \propto E^m \quad (2)$$

where  $E$  is the incident laser intensity,  $m$  is the number of photons involved in the generation of the thermal lens signal,  $I(t=0)$  and  $I(t=\infty)$  are the thermal lens signal strength at  $t=0$  and  $t=\infty$ , respectively. Hence, by monitoring the dependence of the thermal lens signal amplitude on pump intensity, one can identify the occurrence of different processes like one photon absorption ( $m=1$ ) and multiphoton absorption processes ( $m \geq 2$ ). We have measured thermal lens signal produced from solutions of C<sub>60</sub> in toluene at different input energies. Figure 5 gives a typical log-log plot of C<sub>60</sub> in toluene. As is clear from the figure, a linear transmittance of a 30% C<sub>60</sub> toluene solution gives a slope value

The occurrence of slope 2 in these curves is an indication of two-photon absorption (TPA). However, the optical limiting action of C<sub>60</sub> can be explained in terms of transient laser saturable absorption, although some other mechanisms like nonlinear refraction, nonlinear scattering, etc. may play a role under certain experimental conditions. The excitation of instantaneous or simultaneous TPA in C<sub>60</sub> has not yet been demonstrated under laser excitation in the nanosecond range. The two-photon absorption from the triplet state in these molecules is less likely since a considerable triplet absorption cross section is reported at 532 nm.<sup>32</sup> Another potentially effective nonlinearity is sequential two-photon absorption where the excited state absorption cross section is larger than the ground state absorption cross section. The observation of slope 2 in the double logarithmic plot of these molecules could be due to sequential two-

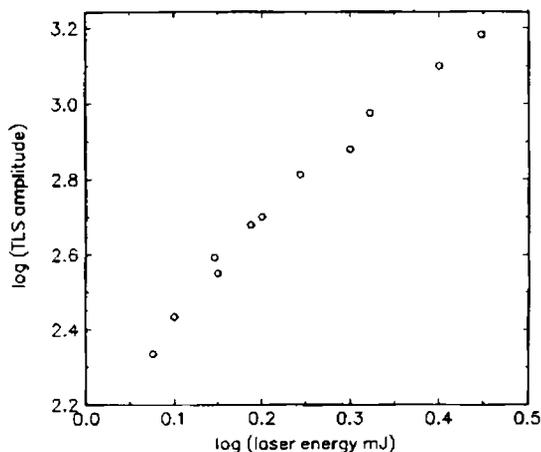


FIG. 5. Log-log plot of thermal lens signal (TLS) amplitude as a function of input laser energy for C<sub>60</sub>.

photon absorption. That is, instead of simultaneous absorption of many photons and exciting the molecule to a higher level which is multiphoton resonant, the photons are absorbed one after the other. In the power limiting region, the leading part of the laser pulse excites most of the molecules to the vibrational levels of the excited singlet state  $S_1$  which rapidly decay nonradiatively to the lowest vibrational level of  $S_1$ . The reported quantum yield for the intersystem crossing rate to the lowest triplet  $T_1$  for C<sub>60</sub> is nearly unity.<sup>23</sup> As the intersystem crossing is very fast for C<sub>60</sub> ( $\ll$  ns) for ns pulses, the triplet state  $T_1$  gets rapidly populated. The triplet state  $T_1$  has a long lifetime (40  $\mu$ s for C<sub>60</sub>) and a higher absorption cross section than that of the ground state. After resonantly absorbing a single photon, the molecule in the triplet state goes to a higher excited state ( $T_n$ ), the lifetime of which is considered to be in the picosecond region due to fast internal conversion. The molecule can relax to the lower excited triplet level by collisional energy transfer to the surrounding solvent molecules. This process occurs within the duration of the nanosecond laser pulse. Thus, the nonradiative decay from  $T_n$  gives rise to appreciable thermal lens signals with a quadratic dependence on laser energy.

For C<sub>60</sub>, where absorption from the excited singlet states is negligible for ns pulses, the variation of light intensity  $I(\lambda)$ , at laser wavelength  $\lambda$ , along the beam propagation direction in a cell containing these solutions will be given, at an instant  $t$  during the laser pulse, by the following expression:<sup>32</sup>

$$\frac{dI(\lambda)}{dz} = -\alpha_0 S_0(t)I(\lambda) - \alpha_1 T_1(t)I(\lambda), \quad (3)$$

where  $S_0(t)$  and  $T_1(t)$  are the ground and triplet state concentrations, respectively, at instant  $t$ , and  $\alpha_0$  and  $\alpha_1$  are the linear absorption coefficients from the  $S_0$  and  $T_1$  states, respectively.

The triplet state concentration at instant  $t$  is, to a good approximation, given by

$$T_1(t) = \alpha_0 S_0(t) I(\lambda) \Phi_{isc}, \quad (4)$$

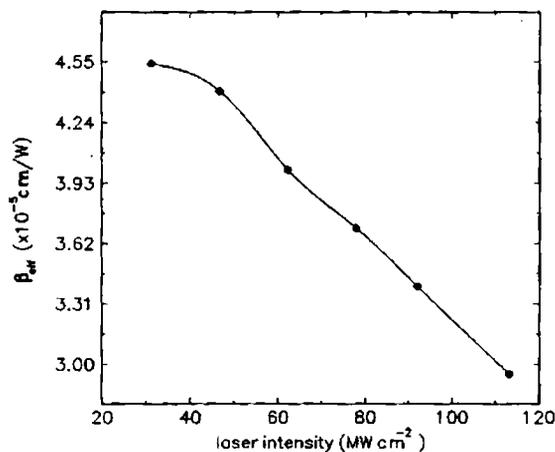


FIG. 6. Effective STPA parameter  $\beta_{\text{eff}}$  as a function of the incident laser intensity for  $C_{60}$ .

where  $\Phi_{\text{isc}}$  is the intersystem crossing yield. For  $C_{60}$ ,  $\Phi_{\text{isc}} \approx 1$ , so that

$$\begin{aligned} \frac{dI(\lambda)}{dz} &= -\alpha_0 S_0(t)I(\lambda) - \alpha_1 \alpha_0 S_0(t)I(\lambda)I(\lambda) \\ &= -\alpha I(\lambda), \end{aligned} \quad (5)$$

where  $\alpha = -\alpha_0 S_0(t) - \alpha_1 \alpha_0 S_0(t)I(\lambda)$  can be considered as an intensity dependent absorption coefficient. This  $\alpha$  represents the absorption coefficient for the sequential two-photon absorption (STPA) with an effective STPA parameter  $\beta_{\text{eff}} = \alpha_1 \alpha_0 S_0(t)$ . The above equation is formally similar to  $\alpha = \alpha_0 + \beta I$  for the case of instantaneous two-photon absorption (ITPA). The quantity  $\beta_{\text{eff}}$  is intensity dependent through the  $S_0(t)$  term, while in the ITPA case,  $\beta$  is constant. The dependence of the effective STPA parameter  $\beta_{\text{eff}}$  as a function of the incident intensity  $I_0$  is depicted in Fig. 6 for  $C_{60}$ . It is noted that the value of  $\beta_{\text{eff}}$  decreases with respect to increasing intensity of radiation. The falloff of  $\beta_{\text{eff}}$  with increasing  $I_0$  is a consequence of STPA. With increasing intensity, the total absorption of the  $C_{60}$  approaches asymptotically the value of  $\alpha_1$ , i.e., the absorbance of the triplet state.<sup>18</sup>

## V. CONCLUSIONS

Optical limiting properties of  $C_{60}$  molecules in toluene are studied using 532 nm, 9 ns pulses from a frequency-doubled Nd:YAG laser. Because of large excited singlet as well as triplet absorption cross sections compared to ground state absorption cross section, the major mechanism for this limiting behavior of these molecules is reverse saturable absorption.

Thermal lensing studies in these solutions show a quadratic dependence of the thermal lens signal amplitude versus input laser energy. The occurrence of instantaneous TPA is less probable in the present context, since we are using ns laser pulses for pumping. Therefore, triplet state absorption

via sequential two-photon absorption is the process behind the observation of slope 2 in the log-log plots. This is supported by the fact that the effective value of the triplet photon absorption cross-section decreases with increasing laser intensity. Hence, thermal lensing studies in fullerenes lead to the conclusion that sequential TPA (RSA), which gives a slope of 2 is playing the leading role in the optical limiting properties of  $C_{60}$ .

## ACKNOWLEDGMENTS

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- <sup>1</sup>H. W. Kroto, J. R. Heath, S. C. O. Brien, R. F. Curl, and R. E. Smalley, *Nature* (London) **318**, 162 (1985).
- <sup>2</sup>S. S. Harilal, C. V. Bindhu, R. C. Issac, V. P. Nampoore, and C. P. G. Vallabhan, *J. Appl. Phys.* **82**, 2140 (1997).
- <sup>3</sup>S. S. Harilal, C. V. Bindhu, V. P. Nampoore, and C. P. G. Vallabhan, *J. Appl. Phys.* **72**, 167 (1998).
- <sup>4</sup>S. S. Harilal, C. V. Bindhu, R. C. Issac, V. P. Nampoore, and C. P. G. Vallabhan, *Jpn. J. Appl. Phys., Part 1* **36**, 134 (1997).
- <sup>5</sup>S. W. McElvany, M. M. Ross, and J. H. Callahan, *Acc. Chem. Res.* **25**, 162 (1992).
- <sup>6</sup>R. E. Smalley, *Acc. Chem. Res.* **25**, 98 (1992).
- <sup>7</sup>S. S. Harilal, R. C. Issac, C. V. Bindhu, V. P. Nampoore, and C. P. G. Vallabhan, *J. Appl. Phys.* **80**, 3561 (1996).
- <sup>8</sup>S. S. Harilal, R. C. Issac, C. V. Bindhu, V. P. Nampoore, and C. P. G. Vallabhan, *J. Appl. Phys.* **81**, 3637 (1997).
- <sup>9</sup>W. Krätschmer, L. D. Lamb, K. Fostiropoulos, and D. R. Huffman, *Nature* (London) **347**, 354 (1990).
- <sup>10</sup>L. W. Tutt and A. Kost, *Nature* (London) **356**, 225 (1992).
- <sup>11</sup>J. N. Arbogast, A. P. Darmanyan, C. S. Foote, Y. Rubin, F. N. Diederich, M. M. Alvarez, S. J. Anz, and R. L. Whetten, *J. Phys. Chem.* **95**, 1000 (1991).
- <sup>12</sup>S. D. Sibley, S. M. Argentine, and A. H. Francis, *Chem. Phys. Lett.* **187** (1992).
- <sup>13</sup>A. Andreoni, M. Bondoni, and G. Consolati, *Phys. Rev. Lett.* **72**, 1874 (1994).
- <sup>14</sup>D. Kim and M. Lee, *J. Am. Chem. Soc.* **114**, 4429 (1992).
- <sup>15</sup>L. W. Tutt and T. F. Boggess, *Prog. Quantum Electron.* **17**, 299 (1993).
- <sup>16</sup>M. P. Joshi, S. R. Mishra, H. S. Rawat, S. C. Mehendale, and S. Rustagi, *Appl. Phys. Lett.* **62**, 1763 (1993).
- <sup>17</sup>S. R. Mishra, H. S. Rawat, M. P. Joshi, and S. C. Mehendale, *Appl. Phys. Lett.* **63**, 223 (1996).
- <sup>18</sup>D. G. McLean, R. L. Sutherland, M. C. Brant, D. M. Brandelik, Fleitz, and T. Pottenger, *Opt. Lett.* **18**, 858 (1993).
- <sup>19</sup>S. R. Mishra, H. S. Rawat, M. P. Joshi, and S. C. Mehendale, *J. Phys. Chem.* **98**, 1157 (1994).
- <sup>20</sup>F. N. Diederich and R. L. Whetten, *Acc. Chem. Res.* **25**, 119 (1992).
- <sup>21</sup>Y. Sun, Q. Gong, S. Yang, Y. H. Zou, L. Fei, X. Zhou, and D. Qian, *Chem. Commun.* **102**, 205 (1993).
- <sup>22</sup>A. Kost, L. W. Tutt, M. B. Klien, T. K. Dougherty, and W. E. Elias, *Opt. Lett.* **18**, 334 (1993).
- <sup>23</sup>J. W. Arbogast and C. S. Foote, *J. Am. Chem. Soc.* **113**, 8886 (1991).
- <sup>24</sup>Y. Kojima, T. Matsuoka, N. Sato, and H. Takahashi, *Macromolecules* **28**, 2893 (1995).
- <sup>25</sup>A. J. Twarowski and D. S. Kliger, *Chem. Phys.* **20**, 253 (1977).
- <sup>26</sup>A. J. Twarowski and D. S. Kliger, *Chem. Phys.* **20**, 259 (1977).
- <sup>27</sup>C. V. Bindhu, S. S. Harilal, V. P. N. Nampoore, and C. P. G. Vallabhan, *Opt. Eng. (Bellingham)* **37**, 2791 (1998).
- <sup>28</sup>C. V. Bindhu, S. S. Harilal, V. P. N. Nampoore, and C. P. G. Vallabhan, *Phys. D* (in press).
- <sup>29</sup>C. V. Bindhu, S. S. Harilal, V. P. N. Nampoore, and C. P. G. Vallabhan, *Mod. Phys. Lett. B* **9**, 1471 (1995).
- <sup>30</sup>C. V. Bindhu, S. S. Harilal, V. P. N. Nampoore, and C. P. G. Vallabhan, *Curr. Sci.* **75**, 167 (1998).
- <sup>31</sup>M. Terazima, N. Hirota, H. Shimohara, and Y. Saito, *J. Phys. Chem.* **95**, 9081 (1991).
- <sup>32</sup>S. Couris, E. Koudoumas, A. A. Ruth, and S. Leach, *J. Phys. B* **26**, 1001 (1995).